#### chemistry 11 course review pdf

Chemistry 11 Course Review Chemistry 11â€"Course Review Page 9 7. A compound was analyzed and the following results were obtained: Molar mass: 270.4 g/mol Mass of sample: 162.24 g Mass of potassium: 46.92 g Mass of sulphur: 38.52 g Mass of oxygen: the remainder of the sample is oxygen a) Determine the mass of oxygen in the sample.

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123.11 g of zinc nitrate, Zn(NO 3)2 are dissolved in enough water to form 650.0 mL of solution. Calculate the [Zn(NO 3 ) 2 ]) Include proper units in your work and in your answers.

### Chemistry 11 â€" Course Review

Page 1 Chemistry 11 â€" Course Review Note: This review is meant for practice purposes. There are multiple outcomes that have not been covered in this booklet, which should be part of your personal review.

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8. 123.11 g of zinc nitrate, Zn(NO 3) 2 are dissolved in enough water to form 650.0 mL of solution. Calculate the [Zn(NO 3) 2]) Include proper units in your work and in your answers. Answer \_\_\_\_\_ 9. Calculate the mass of potassium sulphite (K 2SO 3) needed to make 800.0 mL of a 0.200 M solution of K 2SO 3. Include proper units in your work and in your answers.

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Chemistry 11 Organic Chemistry Review Assignment 1. Identify the following as either an alkane, alkene or alkyne. a. C 4H 8 b. C 4H 10 c. C 14H 26 d. C 40H 78 e. C 25H 52 2. Identify which class of organic compounds each of the following belongs to: halocarbon, alcohol, ether, aldehyde, ketone, carboxylic acid, ester, amine, amide, alkane, alkene, or alkyne.

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Review of the Equilibrium Expression, Keq The equilibrium expression for the generalized reaction is determined experimentally by measuring concentrations for solutions (K c) or partial pressures for gases (K p).

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1.11 Dipole Moments and Molecular Polarity 1.12 Intermolecular Forces and ... 2 CHAPTER 1 A Review of General Chemistry must recognize that reactions occur as a result of the motion of electrons. For example, in the ... This chapter reviews some relevant concepts from your general chemistry course that should be familiar to you. Specii¥cally ...

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